

## THE INHIBITION OF *AZOTOBACTER VINELANDII* TERMINAL OXIDASES BY CYANIDE

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### 1. Introduction

The complex, branched respiratory system of the nitrogen-fixing aerobe *A. vinelandii* contains 3 spectrally-distinct cytochromes which can bind carbon monoxide and may therefore be capable of acting as terminal oxidases, viz cytochromes *o*,  $a_1$ , and  $a_2$  [1, 2]. Carbon monoxide action spectra of respiring whole cells or isolated respiratory membranes of *A. vinelandii* have indicated that cytochrome  $a_2$  (*d*) is the major functional oxidase associated with the oxidation of physiological substrates, whereas oxidation of the high potential, artificial substrates ascorbate-TMPD\* and ascorbate-DCPIP occur via cytochrome oxidases  $a_1$  and *o* respectively [2, 3]. Very recently, however, the ability of cytochromes  $a_1$  and *o* to act as terminal oxidases in *A. vinelandii* has been questioned, mainly on the grounds of their low redox potentials [4].

This paper described the effect of the classical cytochrome oxidase inhibitor, potassium cyanide, on the respiratory activity of *A. vinelandii* membranes oxidising a variety of substrates. The results support the concept that this organism contains three cytochrome oxidases which can be functionally isolated by the use of different electron donors and which can be distinguished by their quantitatively and/or qualitatively different responses to cyanide.

### 2. Materials and methods

*A. vinelandii* was batch-cultured at high aeration on nitrogen-free medium and harvested just past the turnover point into oxygen-limited growth [5]; phosphorylating respiratory membranes were prepared and assayed for oxygen uptake and cytochrome content as described previously [5–7]. The concentration of cytochrome  $a_1$  was calculated from the absorption at approximately 595 nm in the reduced *minus* oxidised difference spectrum using the millimolar extinction coefficient for cytochrome  $a_1$   $mM\epsilon = 4.6$ ; [8]). Protein was assayed by the modified biuret method [9]. Potassium cyanide was freshly prepared for each set of experiments by neutralisation with 1 N acetic acid.

The terminal oxidases were regarded as catalysing bi-substrate reactions involving reducing equivalents and molecular oxygen [10]. Oxygen uptake rates were measured at high (i.e. essentially saturating) oxygen concentrations and the concentration of the reduced substrate was varied either directly (by varying the concentration of the electron mediators TMPD or DCPIP) or indirectly (through the oxidation of several physiological substrates by respiratory chain dehydrogenases of different activity [7, 11, 12].

### 3. Results and discussion

The oxidation of physiological substrates by *A. vinelandii* respiratory membranes (predominantly via cytochrome oxidase  $a_2$  [2, 3] was substantially less sensitive to cyanide than was the oxidation of the

\* Abbreviations: TMPD, tetramethylphenylenediamine;  
DCPIP, dichlorophenolindophenol.

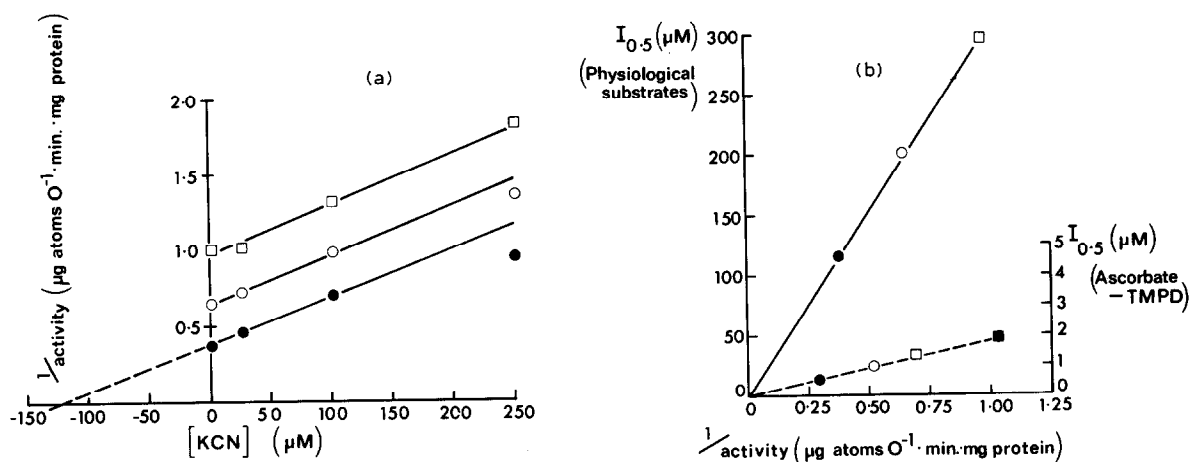


Fig. 1. The effect of cyanide on the oxidation of physiological substrates by *A. vinelandii* respiratory membranes. Oxygen uptake was measured polarographically at 30°C under state 3 conditions as described previously [7]. Respiratory membranes plus reaction mix were incubated with cyanide for 1 min prior to the addition of substrate; initial oxygen uptake rates were recorded: (a) Dixon plot ( $1/v$  versus [cyanide]); substrates, NADH ( $\square$ ); malate ( $\circ$ ) and NADH plus malate ( $\bullet$ ); 0.322 mg protein/2.4 ml assay volume; (b)  $I_{0.5}$  versus  $1/v$  (—) substrates: NADH ( $\square$ ); malate ( $\circ$ ) and NADH plus malate ( $\bullet$ ); data from fig. 1(a). (---) substrate: ascorbate plus varying concentrations of TMPD, 0.44 mM ( $\blacksquare$ ), 0.71 mM ( $\square$ ), 1.77 mM ( $\circ$ ), infinite ( $\bullet$ ); data from fig. 2(b).

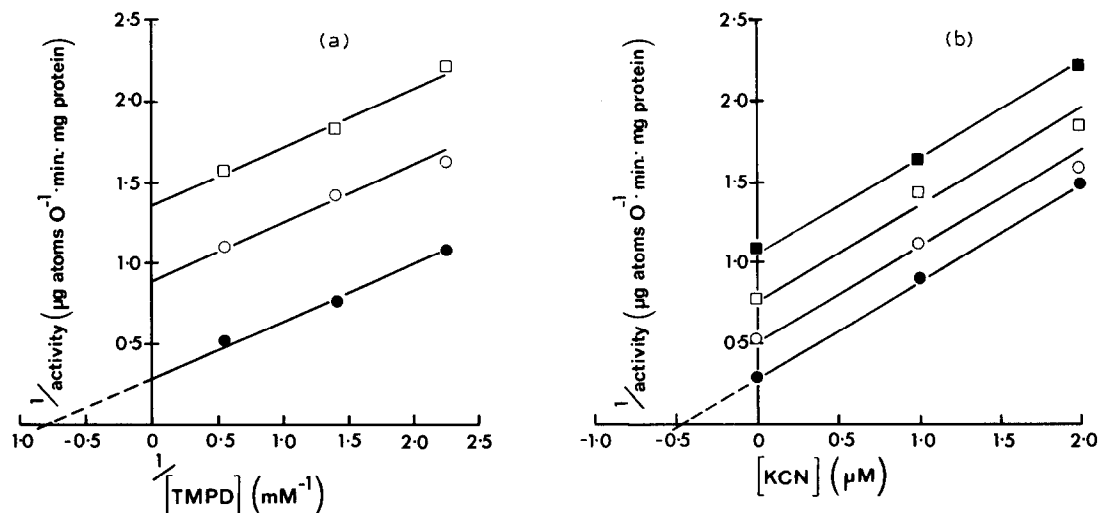


Fig. 2. The effect of cyanide on the oxidation of ascorbate-TMPD by *A. vinelandii* respiratory membranes. Cyanide inhibition of oxygen uptake was measured as for fig 1: (a) Lineweaver-Burk plot ( $1/v$  versus  $1/[\text{TMPD}]$ ) at varying concentrations of cyanide: nil ( $\bullet$ ); 1  $\mu\text{M}$  ( $\circ$ ); 2  $\mu\text{M}$  ( $\square$ ); 0.290 mg protein/2.4 ml assay volume; (b) Dixon plot ( $1/v$  versus [cyanide]) at varying concentrations of TMPD: 0.44 mM ( $\blacksquare$ ); 0.71 mM ( $\square$ ); 1.77 mM ( $\circ$ ); infinite ( $\bullet$ ); 0.290 mg protein/2.4 ml assay volume.

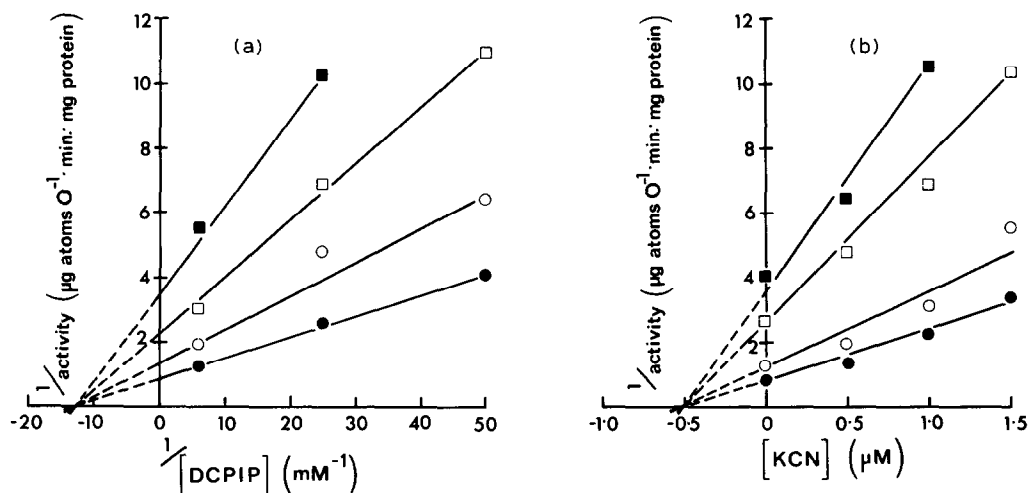


Fig. 3. The effect of cyanide on the oxidation of ascorbate-DCPIP by *A. vinelandii* respiratory membranes. Cyanide inhibition of oxygen uptake was measured as for fig. 1: (a) Lineweaver-Burk plot ( $1/v$  versus  $1/[\text{DCPIP}]$ ) at varying concentrations of cyanide: nil (●); 0.5 μM (○); 1.0 μM (□); 1.5 μM (■); 0.639 mg protein/2.4 ml assay volume; (b) Dixon plot ( $1/v$  versus [cyanide]) at varying concentrations of DCPIP: 20 μM (■); 40 μM (□); 160 μM (○); infinite (●); 0.639 mg protein/2.4 ml assay volume.

artificial, high potential substrates ascorbate-TMPD and ascorbate-DCPIP (via cytochromes  $a_1$  and  $o$  respectively [3]). These results agree with those of earlier investigations [2, 5, 13–15]. Thus, at cyanide concentrations greater than approximately 5 μM, oxygen uptake with NADH, malate or NADH plus malate as substrate, was considered to occur solely via cytochrome  $a_2$ . Under these conditions, linear parallel lines were observed in plots of  $1/v$  versus [cyanide] (fig. 1 (a)) and indicated pure, uncompetitive inhibition of cytochrome  $a_2$ . This was confirmed when  $I_{0.5}$  was plotted against  $1/v$  for each substrate and a straight line which passed through the origin was obtained (fig. 1 (b), solid line). For uncompetitive inhibition  $I_{0.5} = K_i$  only when  $v = V_{\max}$ . Thus, since the observed oxygen uptake rate with NADH + malate as substrate (2.71 μgatom oxygen·min<sup>-1</sup>·mg protein<sup>-1</sup>) probably reflected the maximum rate of electron flux through the central Q-b<sub>1</sub> region of the respiratory chain rather than the  $V_{\max}$  of the oxidase  $a_2$  [7], the  $K_i$  of cyanide for cytochrome  $a_2$  was probably considerably less than the observed minimum  $I_{0.5}$  value of 115 μM. Although the aerobic steady state reduction level of cytochrome  $a_2$  is considerably lower than that of cytochrome  $b_1$  [2], it is difficult to envisage a  $V_{\max}$  for cytochrome  $a_2$  greater than 20 μgatom oxygen·min<sup>-1</sup>·mg protein<sup>-1</sup>

(equivalent to a turnover number of approximately 1900 e·sec<sup>-1</sup>) and the  $K_i$  is therefore unlikely to be less than 15 μM cyanide.

The inhibition by cyanide of ascorbate-TMPD oxidation (cytochrome  $a_1$ ) was also of the pure, uncompetitive type as indicated by the family of linear, parallel lines which were obtained in plots of  $1/v$  versus  $1/[\text{TMPD}]$  (fig. 2 (a)) and of  $1/v$  versus [cyanide] (fig. 2 (b)). This was confirmed when  $I_{0.5}$  was plotted against  $1/v$  and a straight line which passed through the origin was obtained (fig. 1 (b); ———); the slope of this line was considerably less than that obtained during the oxidation of physiological substrates and reflected that much greater sensitivity to cyanide of cytochrome oxidase  $a_1$  than cytochrome  $a_2$ . At an infinite concentration of TMPD the  $V_{\max}$  was 3.50 μgatom oxygen·min<sup>-1</sup>·mg protein<sup>-1</sup> (equivalent to a turnover number of cytochrome  $a_1$  of 865 e·sec<sup>-1</sup>) and the  $K_i$  ( $\equiv I_{0.5}$ ) for cyanide was 0.46 μM, i.e. at least one order of magnitude less than that of cytochrome  $a_2$ . In contrast, cyanide inhibited the oxidation of ascorbate-DCPIP (cytochrome  $o$ ) in a pure, non-competitive manner (fig. 3 (a), (b)). At an infinite concentration of DCPIP the  $V_{\max}$  was 1.16 μgatom oxygen·min<sup>-1</sup>·mg protein<sup>-1</sup> (equivalent to a turnover number for cytochrome  $o$  of 357 e·sec<sup>-1</sup>) and the

$K_i$  for cyanide was  $0.51 \mu\text{M}$ , i.e. almost identical to that of cytochrome  $a_1$ .

The effect of cyanide on the oxidation by *A. vinelandii* respiratory membranes of physiological substrates, ascorbate-TMPD or ascorbate-DCPIP is clearly quantitatively and/or qualitatively different and suggests the presence of three functional terminal oxidases (identified by carbon monoxide action-spectra as cytochromes  $a_2$ ,  $a_1$ , and  $o$  respectively [13]). The pure, uncompetitive inhibition of cytochrome oxidases  $a_1$  and  $a_2$  by cyanide (albeit with widely different  $K_i$  values) suggests that this inhibitor combines principally with ES to yield a stable oxidase-inhibitor complex. However, whether ES represents the reduced [11, 16] or oxygenated [17] forms of the oxidases remains in doubt. Pure uncompetitive inhibition of all three oxidases was observed using carbon monoxide (C.W. Jones, unpublished data), an inhibitor which combines only with the reduced forms. The pure, non-competitive inhibition of cytochrome  $o$  affords little clue to the mode of action of cyanide on this oxidase.

*A. vinelandii* cytochrome oxidases  $o$  and  $a_1$  are clearly much more sensitive than cytochrome  $a_2$  to cyanide, a conclusion which has already been drawn for at least two other  $o$   $a_1$   $a_2$  multiple-oxidase systems [18, 19]. Cyanide inhibition kinetics similar to those found with *A. vinelandii* have also been observed for cytochrome oxidases  $a_1$  and  $o$  from other bacteria (D.J. Meyer and C.W. Jones, unpublished data).

The low redox potentials of *A. vinelandii* cytochromes  $o$  and  $a_1$  [4] obviously offer no barrier to these cytochromes acting as terminal oxidases, but would imply that they might terminate respiratory chains of low oxidative phosphorylation efficiency. There is good evidence to suggest that this may be true for cytochrome oxidase  $a_1$  [5, 20], but there appears to be conflicting evidence in the case of cytochrome  $o$  [5, 20–22].

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